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Article

In Vitro Antimicrobial and Antiproliferative Activities of the Root Bark Extract and Isolated Chemical Constituents of *Zanthoxylum paracanthum* Kokwaro (Rutaceae)

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Abstract: *Zanthoxylum paracanthum* Kokwaro (Rutaceae) is an endemic Kenyan and Tanzanian plant used in folk medicine by local populations. Although other *Zanthoxylum* species have been studied, only *Z. paracanthum* stem extracts have been profiled, even though the roots are also used as herbal remedies. As root extracts may be another source of pharmaceutical compounds, the CH₂Cl₂/MeOH (1:1) root bark extract was studied in this report. Eight root bark compounds were isolated and their structural identities were confirmed by mass spectrometry (MS) and nuclear magnetic resonance (NMR) (using COSY, HSQC, NOESY and HMBC) analyses. The structural identities were determined as follows: the fatty acid—myristic acid (1); the sterol—stigmaterol (2); the lignan—sesamin (3); two β-carboline alkaloids—10-methoxycanthin-6-one (6) and canthin-6-one (7); and three phenanthridine alkaloids—8-acetonyldihydrochelerythrine (4), arnottianamide (5) and 8-oxochelerythrine (8). Some of these compounds were identified in the species for the first time. These compounds and the extract were then tested in vitro against methicillin-resistant *Staphylococcus aureus* (MRSA), *Escherichia coli* (ATCC 25922), *Staphylococcus aureus* (ATCC 29213) and *Candida albicans* (ATCC 10231) before tests for antiproliferative activity against the human breast cancer (HCC 1395), human prostate cancer (DU 145) and normal (Vero E6) cell lines were conducted. Minimum inhibition concentration values of 3.91, 1.95, 0.98 and 7.81 µg/mL against MRSA, *S. aureus*, *E. coli* and *C. albicans*, respectively, were recorded. Among the isolates, canthin-6-one was the most active, followed by 10-methoxycanthin-6-one. The root extract and some of the compounds also had antiproliferative activity against the HCC 1395 cell line. Stigmaterol and canthin-6-one had IC₅₀ values of 7.2 and 0.42. The root bark extract also showed activity, at 8.12 µg/mL, against the HCC 1395 cells. Out of the chemical isolates, 10-methoxycanthin-6-one and canthin-6-one showed the strongest inhibition of the DU 145 cells. The root extract had significant antimicrobial and antiproliferative activities, supporting the traditional use of this plant in treating microbial infections and cancer-related ailments.

Keywords: alkaloids; antimicrobial activity; anticancer activity; canthinones; medicinal plant; phytochemistry; traditional medicine

1. Introduction

Medicinal plants remain an important source of pharmaceuticals, and in communities where access to healthcare is limited, plant-based medications have a high proportion of use [1]. In Africa, including Kenya, the use of herbal medicines to treat different ailments, including infectious diseases and cancers, is common [2]. Microbial diseases are often implicated in the onset of cancers, and compounds that may have both antimicrobial and anticancer effects are thus sought after. In Kenya, breast, cervical, oesophageal, colon, stomach and ovarian cancers occur at higher incidents in women, while in men, prostate, oesophageal, colon, stomach, oral and liver cancers are most prevalent [3]. With urbanisation, higher levels of cancers are becoming more apparent, and a recent study in 2019 showed that there are more female cancer cases than male; 60% and 56.4%, respectively [4]. There is thus a global commitment to finding new drug entities that are effective in reducing the proliferation of cancers [5]. In the search for new plant-derived pharmaceuticals, the candidate drug should demonstrate selectivity by having low levels of toxicity to normal cell lines, and superior cytotoxicity towards cancer cell lines [6]. In general, natural products with pharmacological effects can become lead structures in the preparation of semi-synthetic and synthetic drugs [7]. In this regard, the plants used in traditional medicine remain sources of such structures [8]. Incidentally, a large fraction of medicinal plants lack the scientific backup to substantiate their traditional uses [1]. Additionally, their toxicities and safe dosages are unknown [9]. *Zanthoxylum paracanthum* Kokwaro (Family Rutaceae) is a shrub or a small tree occurring in the coastal areas of Kenya and Tanzania, and it has a long history as a traditional medicine in the areas where it is found [10]. Although some of the other *Zanthoxylum* species are quite well studied, little ethnobotanical information exists because of their narrow distribution. Communities living in the coastal areas of Kenya use stem and root extracts of this plant in the management of tumors and other related diseases [11], but thus far, only stem extracts have been chemically profiled. In Tanzania, the leaves are used as an indigenous vegetable, as they have a high nutrient content and it is claimed they cure gastrointestinal problems such as diarrhoea [12], suggesting the antimicrobial potential of the plant. The isolation of three new alkaloids, zanthoxoaporphines A–C with larvicidal activity against the third-instar *Anopheles gambiae* larvae [13], from a *Z. paracanthum* stem extract has shown the possibility of discovering new compounds from this species. In another study, canthin-6-one, isolated from the stem bark extract of *Z. paracanthum*, was found to be active against both drug-sensitive and multidrug-resistant leukemia cell lines [11].

As there are close to 250 *Zanthoxylum* species, only some plants from this taxon have been investigated for their chemical and pharmacological properties. The antimicrobial and cytotoxicity effects have been studied in species such as *Z. budruga* [14], *Z. chalybeum* [15], *Z. leprieurii* and *Z. zanthoxyloides* [16]. As the root bark extract of *Z. paracanthum* has never been studied in this way before, we identified this extract as a possible new source of bioactive compounds. Eight compounds were purified from the root bark extract, and here, the in vitro antimicrobial and antiproliferative activities of the root bark extract and the eight chemical isolates of *Z. paracanthum* are reported.

2. Results and Discussion

2.1. Identification of the Isolated Compounds

Eight known compounds were isolated (Figure 1) and identified, by MS and NMR (using COSY, HSQC, NOESY and HMBC) analyses (see Supplementary Materials) as myristic acid (1) [17], stigmasterol (2) [18], sesamin (3) [11], 8-acetyldihydrochelerythrine (4) [19], arnottianamide (5) [20], 10-methoxycanthin-6-one (6) [21], canthin-6-one (7) [11] and 8-oxochelerythrine (8) [22].

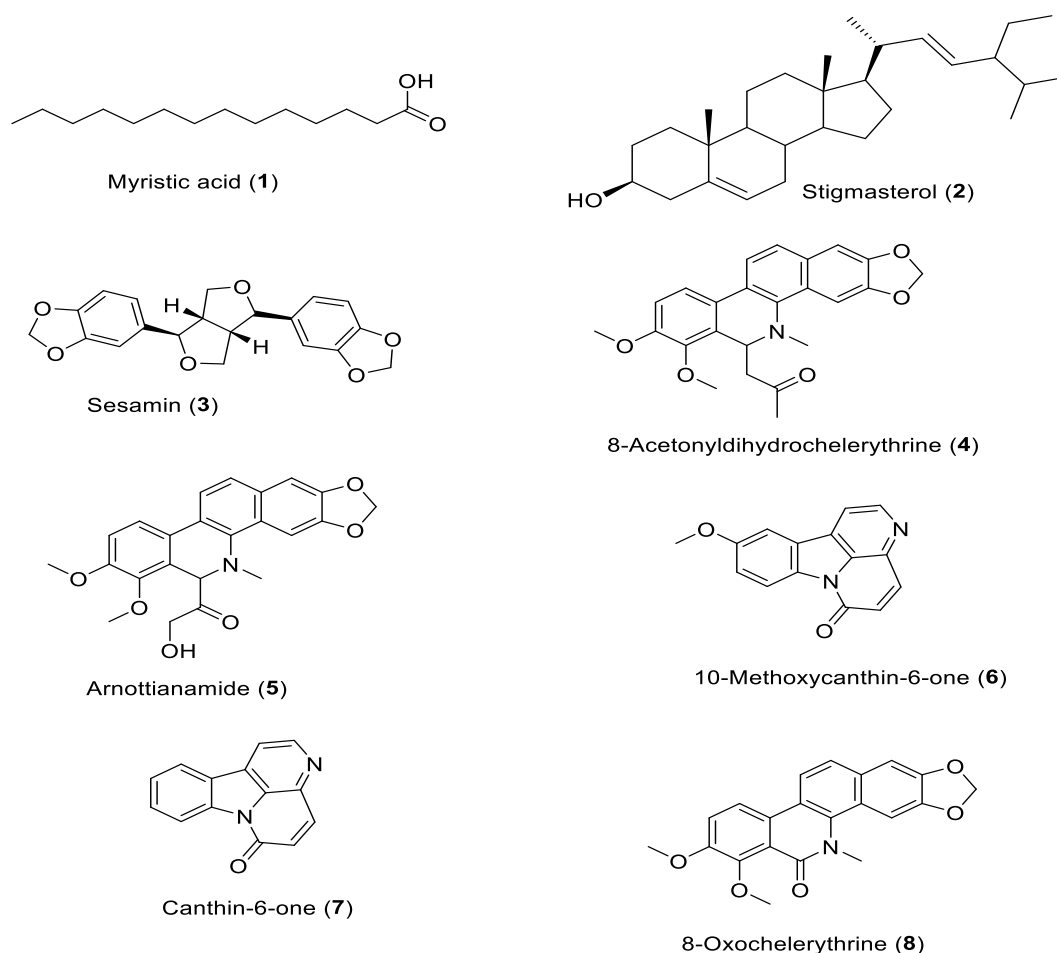


Figure 1. Structures of compounds (1–8) isolated from root bark of *Zanthoxylum paracanthum*.

2.2. Antimicrobial Activity

The $\text{CH}_2\text{Cl}_2/\text{CH}_3\text{OH}$ (1:1) extract from the root bark of *Z. paracanthum* had high antimicrobial activity against all the tested microbes (Table 1). The minimal inhibition concentration (MIC) values of this extract were 3.91, 1.95, 0.98 and 7.81 $\mu\text{g}/\text{mL}$, against MRSA, *E. coli*, *S. aureus* and *C. albicans*, respectively. This is the first report on the antimicrobial activity of *Z. paracanthum*. The activity level observed here is similar to the antimicrobial activities reported for other Kenyan *Zanthoxylum* species. *Z. chalybeum* elicited a broad-spectrum antimicrobial activity against *Bacillus cereus*, *Pseudomonas aeruginosa*, MRSA and *Candida albicans* [23]. The high and broad-spectrum antimicrobial activities of *Z. gillettii* and *Z. holstzianum* have been reported [24,25]. These studies support the traditional use of *Zanthoxylum* species in Kenya in the broad-spectrum treatment of microbial infections, and because the extract exhibited good and promising antimicrobial effects, the compounds that exhibit such activity were of interest.

The eight isolated phytochemicals were tested for antimicrobial activity, of which myristic acid (1) had no activity against all the tested microbes with MIC values of more than 1000 $\mu\text{g}/\text{mL}$ (Table 1). This is in line with the previous antimicrobial studies on this saturated fatty acid, myristic acid, which recorded no activity against MRSA [26]. Plant sterols are often a good source of antimicrobial agents, and in this study stigmasterol (2) obtained from the root bark extract did not inhibit the growth of MRSA and *C. albicans*, but showed moderate activity against the growth of *S. aureus* and *E. coli*, with MIC values of 62.50 $\mu\text{g}/\text{mL}$ and 15.63 $\mu\text{g}/\text{mL}$, respectively. Elsewhere, a study on the antimicrobial activity of stigmasterol indicated its activity mainly against *Streptococcus mutans* and *S. sobrinus* strains, but with very limited activity [27]. Sesamin (3) is a lignan, and occurs in a wide variety of plant species, including *Sesamum indicum* L. [28]. Sesamin had no activity against MRSA, *E. coli* or *C. albicans*, but had

an inhibitory effect against the growth of *S. aureus*, with an MIC value of 500 µg/mL. Previous studies on the antimicrobial and antioxidant properties of sesamin revealed low antibacterial activity [29]. Additionally, sesamin did not inhibit the growth of MRSA, the resistant strain of *S. aureus* [30].

Table 1. Antimicrobial minimal inhibition concentration (MIC) values (µg/mL) of the tested samples.

Sample Name	Microbial Organism			
	MRSA	<i>S. aureus</i>	<i>E. coli</i>	<i>C. albicans</i>
myristic acid (1)	>1000	>1000	>1000	>1000
stigmaterol (2)	>1000	62.50	15.63	>1000
sesamin (3)	>1000	500	>1000	>1000
8-acetyldihydrochelerythrine (4)	31.25	15.63	15.63	62.50
arnottianamide (5)	>1000	>1000	>1000	>1000
10-methoxycanthin-6-one (6)	3.91	1.95	3.91	7.81
canthin-6-one (7)	0.98	0.49	1.95	3.91
8-oxochelerythrine (8)	62.50	7.81	3.91	15.63
<i>Z. paracanthum</i> root bark extract	3.91	0.98	1.95	7.81
Omacilin	0.98	0.49	0.98	-
Fluconazole	-	-	-	1.95

MIC—Minimal inhibition concentration. MRSA—methicillin resistant *Staphylococcus aureus*, *Escherichia coli* (ATCC 25922), *Staphylococcus aureus* (ATCC 29213) and *Candida albicans* (ATCC 10231). MIC values for canthin-6-one (7) against MRSA and *S. aureus* were similar to omacilin (positive control).

One of the most potent compounds isolated in this study is 8-acetyldihydrochelerythrine (4), which inhibited the growth of all four microbes, with MIC values of 31.25 µg/mL against MRSA, 15.63 µg/mL against *S. aureus* and *E. coli*, and 62.50 µg/mL against *C. albicans*. These results are in agreement with what has been reported for the same compound isolated from *Zanthoxylum rhetsa* (Roxb.) DC. roots, which exhibited strong activity against MRSA (MIC = 8 µg/mL) and moderate activity against *E. coli* (MIC = 16 µg/mL) [31]. Arnottianamide (5) was among the isolated compounds that did not inhibit the growth of all the microbes tested (MIC above 1000 µg/mL), which is in agreement with the reports that indicated that this compound did not exhibit any antimicrobial activity against the microbes tested [32,33].

Zanthoxylum species are regarded as good sources of alkaloids, and here we have isolated and determined the antimicrobial effects of the alkaloid 10-methoxycanthin-6-one (6). This compound possessed antimicrobial activity against all the microorganisms tested. The MIC values of this compound against each microbe were 3.91 µg/mL against MRSA, 1.95 µg/mL against *S. aureus*, 3.91 µg/mL against *E. coli* and 7.81 µg/mL against *C. albicans*. Previous studies illustrate that this compound exhibited high inhibitory activities against *Bacillus cereus*, *Bacillus subtilis*, *Ralstonia solanacearum* and *Pseudomonas syringae*, with MIC values between 3.91 and 31.25 µg/mL, and significant antifungal activity against *Fusarium graminearum* [34]. This is the first report on the antimicrobial activity of this compound against the four tested microbes. A study on structure–activity relationship (SAR) showed that the high antimicrobial activity is conferred by the aliphatic ester derivatives of 10-hydroxycanthin-6-one [35]. The related alkaloid, canthin-6-one (7), exhibited the highest antimicrobial activity (Table 1) against all of the tested microbes, with activity levels similar to those of the positive controls ($p \geq 0.05$). The recorded MIC values for this compound were 0.97 µg/mL for MRSA, 0.48 µg/mL for *S. aureus*, 1.95 µg/mL for *E. coli* and 3.91 µg/mL against *C. albicans*. Soriano-Agatón et al. [36] reported the antifungal activity of four naturally occurring canthin-6-one derivatives against five pathogenic fungi. However, the structure–activity relationship (SAR) for the antifungal activity remained unclear. A high antimicrobial activity of canthin-6-one against *Staphylococcus aureus* and *Saccharomyces cerevisiae* has also been reported [37].

Finally, the alkaloid 8-oxochelerythrine (8) had significant antimicrobial activity against all the tested microbes, with MIC values of 62.5 µg/mL against MRSA, 31.25 µg/mL against *S. aureus*,

3.91 µg/mL against *E. coli* and 15.63 µg/mL against *C. albicans*. 8-Oxochelerythrine had considerable inhibitory activity against *Clostridium sporogenes* (MIC 0.91 µM) and *Streptococcus pyogenes* (MIC 3.64 µM) [38]. The antimicrobial activity of 8-oxochelerythrine is attributed to the presence of two methoxy groups at C-7 and C-8 of this compound [39].

2.3. Antiproliferative Activity

The antiproliferative activities of the samples were determined and the IC₅₀ values are given in Table 2. The criterion for reporting antiproliferative activity for natural products was based on in vitro cytotoxicity, after the exposure of the cells to the samples for 72 h, as recommended by the U.S. National Cancer Institute (NCI). An IC₅₀ value lower than 20 µg/mL for crude extracts, and an IC₅₀ value less than 4 µg/mL for pure compounds, is considered to be highly antiproliferative [40]. The selective inhibitory activity of the samples was determined and expressed as the selectivity index (SI). The SI values demonstrate the ability of samples to kill cancerous cells without significantly affecting the normal cells. A high SI value depicts high selectivity. Natural products with SI ≥ 2 are considered to be highly selective, while SI < 2 indicates less selectivity [41]. A normal cell line, Vero E6, widely used in toxicology, virology and pharmacology research and in the testing of new vaccines and diagnostic reagents in industry [42], was used in this study. In particular, these cells have been used as models to evaluate the toxicity of compounds of different natures, either chemical or microbial toxins, and are routinely used as a part of screening programs [43].

Table 2. CC₅₀/ IC₅₀ (µg/mL) and SI values for different samples against selected cell lines.

Sample Names	CC ₅₀ Normal Cell Line	IC ₅₀ Cancerous Cell Lines		Selectivity Indexes	
	Vero E6	HCC 1395	DU 145	HCC 1395	DU 145
myristic acid (1)	64.86 ± 0.51 ^c	57.71 ± 1.2 ^a	80.24 ± 0.12 ^d	1.12	0.81
stigmaterol (2)	123.88 ± 0.00 ^b	0.42 ± 0.1 ⁱ	140.49 ± 1.27 ^a	294.94	0.88
sesamin (3)	135.31 ± 0.12 ^a	3.39 ± 1.0 ^h	115.06 ± 0.03 ^b	39.97	1.18
8-acetyldihydrochelerythrine (4)	47.83 ± 1.15 ^e	9.99 ± 0.6 ^e	66.82 ± 0.58 ^e	4.79	0.72
arnottianamide (5)	2.77 ± 0.12 ^h	38.34 ± 0.1 ^b	84.31 ± 0.64 ^c	0.07	0.03
10-methoxycanthin-6-one (6)	53.95 ± 0.38 ^d	14.70 ± 0.5 ^c	1.58 ± 0.00 ⁱ	3.67	34.15
canthin-6-one (7)	41.81 ± 0.64 ^f	8.12 ± 0.6 ^f	9.43 ± 0.01 ^h	5.15	4.43
8-oxochelerythrine (8)	135.32 ± 0.12 ^a	14.09 ± 0.3 ^d	63.41 ± 1.10 ^f	9.60	2.13
<i>Z. paracanthum</i> root bark extract	28.28 ± 0.34 ^g	7.27 ± 0.0 ^g	53.21 ± 1.21 ^g	3.89	0.53
Doxorubicin (Positive control)	0.30 ± 0.12 ⁱ	0.21 ± 0.2 ⁱ	0.59 ± 0.01 ⁱ	1.41	0.51

Vero E6—Normal cell line, HCC 1395—human breast cancer cell line, DU 145—human prostate cancer cell line, SI—selectivity index. Values for CC₅₀/ IC₅₀ are expressed as Mean ± SEM, values sharing a superscript letter in each column are not significantly different from each other ($p \geq 0.05$).

All the compounds exhibited concentration-dependent cytotoxicity levels (Figures 2–4). The antiproliferation effects show a good correlation with the concentrations supplied to the different cell types. Some samples were highly effective against the cancer cells, with high inhibition percentages (above 70%) being recorded (Figures 2A and 3A). The dose dependent curve is biphasic, with a sharp (exponential) linear relationship that is more evident as increasing concentrations of the extract or compounds are included in the growth medium. The higher concentrations of the test samples appear to be correlated with the asymptotic slope of the curve. The effect of lower concentrations, below 4 µg/mL for tested samples, is more clearly shown by Figures 2B, 3B and 4B. In some cases, the root bark extract and some of the compounds were highly active at these low concentrations.

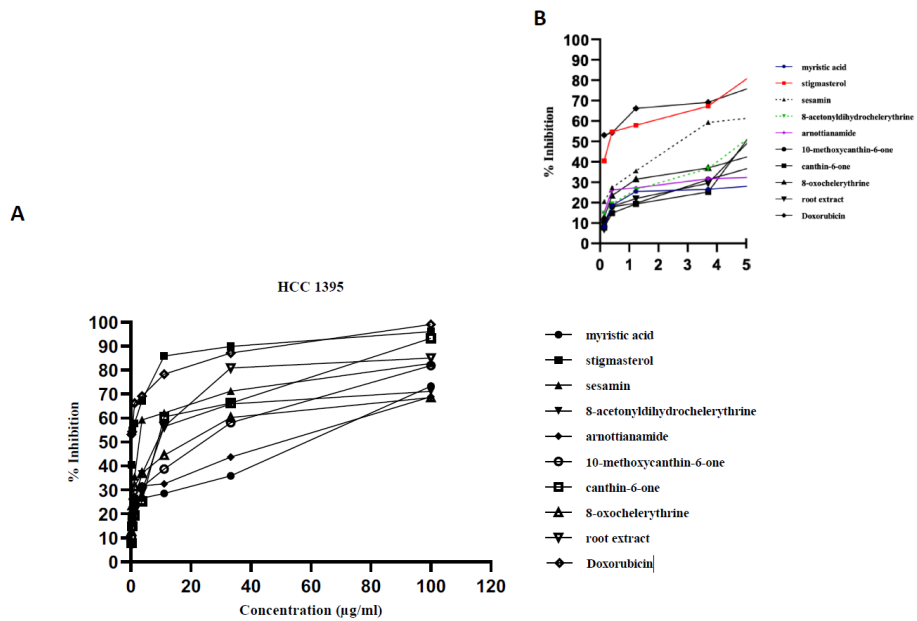


Figure 2. Antiproliferative activity of the tested samples against the human breast cancer cell line (HCC1395). Antiproliferative activity of the tested samples against various cancer cell lines recorded as % growth inhibition. (A) Biphasic growth curve of dose-dependent anti-proliferation effects: 0–100 µg/mL. (B) Effect of test samples at low concentrations: 0–5 µg/mL.

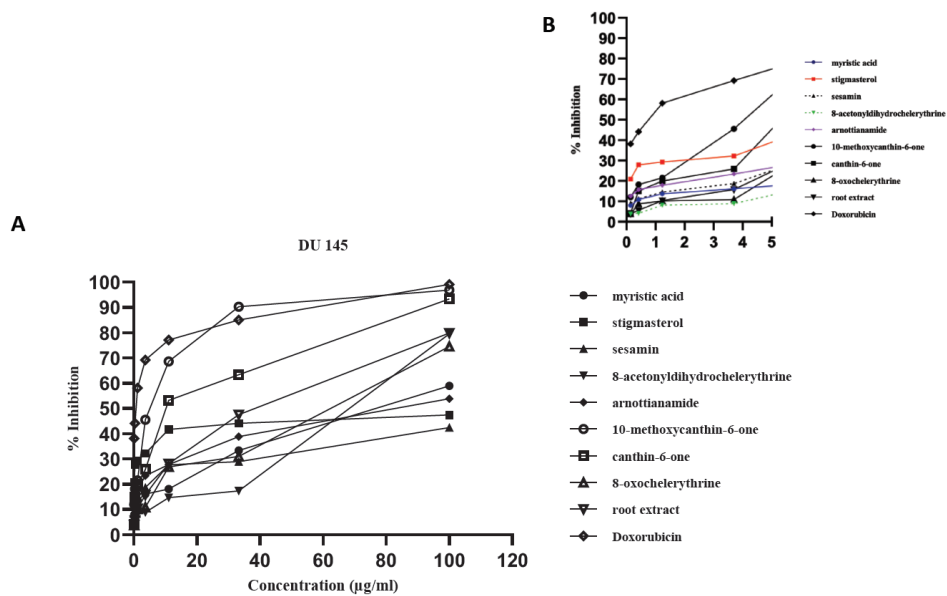


Figure 3. Antiproliferative activity of the tested samples against the human prostate cancer cell line (DU 145). Antiproliferative activity of the tested samples against various cancer cell lines recorded as % growth inhibition. (A) Biphasic growth curve of dose-dependent anti-proliferation effects: 0–100 µg/mL. (B) Effect of test samples at low concentrations: 0–5 µg/mL.

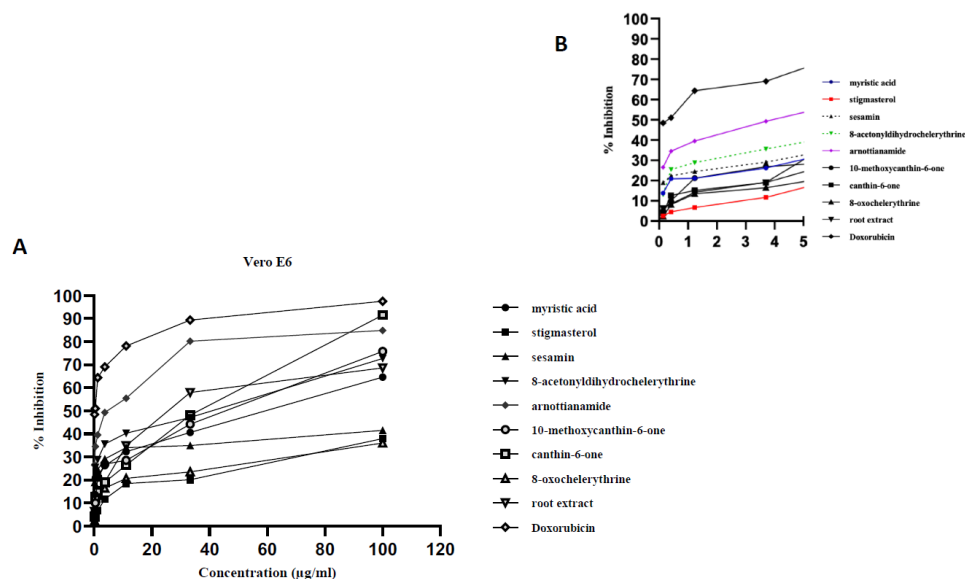


Figure 4. Antiproliferative activity of the tested samples against the normal cell line (Vero E6). Antiproliferative activity of the tested samples against various cancer cell lines recorded as % growth inhibition. (A) Biphasic growth curve of dose-dependent anti-proliferation effects: 0–100 µg/mL. (B) Effect of test samples at low concentrations: 0–5 µg/mL.

The root bark extract of *Z. paracanthum* was highly antiproliferative and selective against the HCC 1395 cell line, with an IC_{50} value of 7.27 ± 0.0 µg/mL, and the SI value was recorded at 3.89 (Table 2). The extract was found to be toxic to normal cells (Vero E6), whilst it moderately inhibited the growth of DU 145 cancer cells with an IC_{50} of 53.21 ± 1.2 µg/mL, and was less selective, with an SI value of 0.51. This is the first report on the antimicrobial and anticancer activity of the root bark extract of *Z. paracanthum*.

The pharmacological effects of the root bark extract of *Z. paracanthum* in this study could be due to the synergistic interactions of the different phytochemicals present in the extract [44] that function on the cell, leading to growth inhibition and even perhaps the apoptosis of microbial and cancer cells, respectively [45]. Studies carried out on the leaf extract of *Zanthoxylum armatum* DC indicated a high anticancer activity against the human cervical cell line [46], while in vitro bioassays of *Z. lepreurii* and *Z. zanthoxyloides* exhibited moderate anticancer and antimicrobial activities [16].

A moderate inhibition of cancer cells was observed with myristic acid (1) against the tested cell lines. The CC_{50} value for Vero E6 was 64.86 ± 0.51 µg/mL, while the IC_{50} values for HCC 1395 and DU 145 were 57.71 ± 1.2 µg/mL and 80.24 ± 0.12 µg/mL, respectively. The selectivity index for the human breast cancer cell line, HCC 1395, was 1.12, while that of the human prostate cancer cell line, DU 145, was 0.81. Myristic acid was therefore less selective against the tested cancer cell lines used in this study (SI values were ≤ 2). This was also observed previously in a study that investigated the effects of branching on fatty acid biosynthesis of human breast cancer cells, wherein myristic acid was found to be cytotoxic to the breast cancer cells [47].

The activity of stigmasterol (2) in the anticancer assays (Table 2) is not surprising. It is interestingly to note that, overall, the potency of this compound in preventing cancer growth was greatest in relation to the human breast cancer cell line (HCC1395). This compound was also effective in inhibiting the growth of the other cancer cell lines, with the human prostate cancer cell line (DU 145) showing a reduced response as compared to the breast cancer cell line, but it was less active against Vero E6. The CC_{50} value for E6 was 123.88 ± 0.00 µg/mL, while the IC_{50} values for HCC 1395 and DU 145 were 0.42 ± 0.1 µg/mL and 140.49 ± 1.27 µg/mL, respectively. The selectivity index for the human breast cancer cell line, HCC 1395, was 294.94. A such, stigmasterol was highly selective against HCC 1395, and less selective against the human prostate cancer cell line, DU 145, where the SI value was 0.88.

This compound is a good candidate for further research on the control of breast cancer, as it has a very low IC_{50} value against HCC1395 as well as a very high selectivity index [48]. In a previous study, the antitumor activity of stigmaterol was evaluated against Ehrlich Ascites Carcinoma (EAC) in swiss albino mice, and the compound was able to decrease the tumor volume and improve the life span of the mice [49].

Lignans have become fascinating as lead compounds with anticancer activity, and there is thus increased interest in them from natural product scientists [50,51]. The lignan, sesamin (3), lacked significant antiproliferative activity against the normal cells Vero E6 (with a CC_{50} value of $135.31 \pm 0.12 \mu\text{g/mL}$) and DU 145 (with an IC_{50} of $115.06 \pm 0.03 \mu\text{g/mL}$). On the other hand, sesamin was highly antiproliferative against HCC 1395, with an IC_{50} value of $3.39 \pm 1.0 \mu\text{g/mL}$. The selectivity index for the human breast cancer cell line, HCC 1395, was 39.97, showing that it was highly selective. However, it was less selective against the human prostate cancer cell line DU 145, with an SI value of 1.18. This study is in agreement with reports that sesamin has a strong chemopreventative activity against mammary tumors in rats [52]. Sesamin was also reported to suppress cell cycle progression and angiogenesis in different mammary cancer models [53,54].

A variable antiproliferative activity was observed for 8-acetyldihydrochelerythrine (4), depending on the cell line used. The CC_{50} value with relation to Vero E6 was $47.83 \pm 1.15 \mu\text{g/mL}$, and the IC_{50} value for DU 145 was $66.82 \pm 0.58 \mu\text{g/mL}$, while the IC_{50} value for HCC 1395 was $9.99 \pm 0.6 \mu\text{g/mL}$. 8-Acetyldihydrochelerythrine was highly selective against HCC 1395, with the SI value of 4.79, and was less selective against DU 145, with which the SI value was 0.72. This is the first report on the antiproliferative activity of 8-acetyldihydrochelerythrine. However, the antiproliferative activities of related phenanthridine alkaloids have been reported, which illustrates that this class of compounds enact significant antitumor activities in cancer cells [55,56].

The antiproliferation effect against Vero E6 was high, with a CC_{50} value of $2.77 \pm 0.12 \mu\text{g/mL}$ being recorded when arnottianamide (5) was applied to the cells. There was also evidence of the moderate antiproliferative activity of this compound against both HCC 1395 and DU 145, with which the IC_{50} values were $38.34 \pm 0.1 \mu\text{g/mL}$ and $84.31 \pm 0.64 \mu\text{g/mL}$, respectively. From the SI values, it was evident that arnottianamide is less selective against both HCC 1395 (with the SI value of 0.07) and DU 145 (with the SI value of 0.03). The lack of specificity of this compound in different cancer cell lines has been reported [57,58].

Other alkaloids that occur in the *Zanthoxylum* species may also be important for the anticancer effects of their extracts. In fact, several compounds that are alkaloids are now in routine use as anticancer drugs, such as taxol [59]. In this study, 10-methoxycanthin-6-one (6) indicated a moderate activity against Vero E6, with a CC_{50} value of $53.95 \pm 0.38 \mu\text{g/mL}$, and an antiproliferative activity against HCC 1395 (IC_{50} value of $14.70 \pm 0.5 \mu\text{g/mL}$) and high antiproliferative activity against DU 145 (IC_{50} value of $1.58 \pm 0.00 \mu\text{g/mL}$). 10-Methoxycanthin-6-one was highly selective against the two cancer cell lines, with an SI values in relation to HCC 1395 of 3.67, while that for DU 145 was 34.15 (Table 2). This is the first report on the antiproliferative activity of 10-methoxycanthin-6-one. Another alkaloid with high anticancer activity is canthin-6-one (7), which, based on the median inhibitory concentration (IC_{50}/CC_{50}) values, was found to have a moderate activity against Vero E6 (with a CC_{50} of $41.81 \pm 0.64 \mu\text{g/mL}$) and an activity against both HCC 1395 and DU 145 (IC_{50} values of $8.12 \pm 0.6 \mu\text{g/mL}$ and $9.43 \pm 0.01 \mu\text{g/mL}$, respectively). The SI values indicated that the compound was highly selective against both HCC 1395 and DU 145, with SI values of 5.15 and 4.43, respectively. A report by Dai et al. [60] indicates that canthin-6-one has a broad anticancer activity, which has made the compound useful as an anticancer agent [61]. Additionally, canthin-6-one isolated from the stem bark of *Z. paracanthum* exhibited a high antiproliferative activity against leukemia cancer cell lines [11].

8-Oxocheleerythrine (8) is shown here for the first time to be toxic against HCC 1395 (with an IC_{50} value of $14.09 \pm 0.3 \mu\text{g/mL}$) and moderately toxic against DU 145 (with an IC_{50} of $63.41 \pm 1.10 \mu\text{g/mL}$). Conversely, it is inactive against the normal cell line, the Vero E6 cell line (a CC_{50} value of $135.32 \pm 0.12 \mu\text{g/mL}$). This compound was highly selective against the human breast cancer cell

line, HCC 1395, with an SI value of 9.60, while that of the human prostate cancer cell line, DU 145, was 2.13. This is the first report on the antiproliferative activity of 8-oxochelerythrine; however, the studies on the antiproliferative activity of some of the other benzophenanthridine alkaloids, already mentioned in this study, authenticate the present report. The percentage inhibitions of the tested samples against HCC 1395, DU 145 and Vero E6 are summarized in Figures 2–4, respectively. The CC_{50}/IC_{50} values and the selectivity indexes for different compounds against selected cell lines are presented in Table 2.

3. Materials and Methods

3.1. General

Column chromatography was done on silica gel 60 (70 230 mesh), and thin layer chromatography (TLC) on silica gel 60 F254, Merck). The NMR spectra were recorded using Bruker Avance 500 MHz spectrometers. COSY, HSQC, NOESY and HMBC spectra were obtained using standard Bruker software. Chemical shifts were measured in ppm in δ values relative to the internal standard tetramethyl silane (TMS). Omacilin and fluconazole were bought from Sigma-Aldrich (USA). The cell lines used in the cytotoxicity tests were obtained from Manassas, VA, USA. Doxorubicin was obtained from Johnson & Johnson, USA. The microorganisms used in this study were provided by the Centre for Microbiology Research (CMR), Kenya Medical Research Institute (KEMRI).

3.2. Plant Materials

The roots of *Z. paracanthum* were collected from Mrima Hills, Kwale County in Kenya (GPS coordinates 4°29'01.6" S and 39°15'30.8" E) in April 2018. The plant was positively identified by Ms Magrate Kaigongi, a plant taxonomist at the Kenya Forestry Research Institute, and a voucher specimen was deposited at the University Nairobi Herbarium (NAI) under voucher specimen number MK14/2018. The plant samples (2 kg) were washed in running water to remove soil, chopped into small pieces (~2 cm) and air dried for three weeks before being ground into fine material using an electrically powered grinder.

3.3. Extraction, Isolation and Elucidation of Compounds

The ground root bark (960 g) of *Z. paracanthum* was exhaustively extracted for 72 h with dichloromethane:Methanol ($CH_2Cl_2:CH_3OH$) (1:1). The resultant extract was filtered and the solvent removed in vacuo to give 128 g of the crude extract. A portion of this extract (77 g) was partitioned between CH_2Cl_2 and water, followed by ethyl acetate (EtOAc) and water. Removal of the organic solvents yielded 45 g of CH_2Cl_2 and 3 g of EtOAc extracts. The CH_2Cl_2 extract (45 g) was subjected to column chromatography on silica gel (700 g) and eluted with *n*-hexane containing increasing amounts of EtOAc (1%, 2%, 4%, 6%, 8%, 10%, 15%, 20%, 30%, ... 90% and 100%) and then with EtOAc containing increasing amounts of CH_3OH (5%, 10%, 20%, ... 100%). A total of 220 fractions, each 400 mL, were collected and combined on the basis of thin layer chromatography (TLC) analysis. The fractions eluted with 1% to 3% EtOAc in *n*-hexane were not followed further. The fractions eluted with 4% EtOAc in *n*-hexane were combined and further purified by column chromatography on Sephadex LH-20 (eluent: CH_2Cl_2/CH_3OH , 1:1) to yield compound 1 (50 mg). The fractions eluted at 6–10% EtOAc in *n*-hexane were combined, then separated by column chromatography over silica gel eluting with *n*-hexane (EtOAc, 4:1) to yield compound 2 (100 mg), compound 3 (320 mg) and compound 4 (101 mg). The fractions obtained at 12% EtOAc were combined and crystallized to give compound 5 (42 mg). The eluents obtained with 15–70% EtOAc in *n*-hexane were combined and purified on Sephadex (eluent: CH_2Cl_2/CH_3OH , 1:1) to afford compound 6 (68 mg), compound 7 (10 g) and compound 8 (54 mg). The EtOAc extract (3 g) was not followed further. The structures of the purified compounds were determined using nuclear magnetic resonance (NMR), spectroscopy, mass spectrometry and comparison with the literature.

3.4. In Vitro Antimicrobial Testing

All the isolated compounds and the root bark extract of *Z. paracanthum* were tested for antimicrobial activity against an un-typed isolate of methicillin resistant *Staphylococcus aureus* (MRSA), *Escherichia coli* (ATCC 25922), *Staphylococcus aureus* (ATCC 29213) and *Candida albicans* (ATCC 10231).

A broth dilution with two-fold serial dilutions of each sample was carried out ranging from 1000 µg/mL to 0.011 µg/mL in 96 micro well plates, to determine the MIC. All the dilutions were carried out in triplicate. One milliliter of 24 h and 72 h culture, for bacteria and fungi respectively ($\cong 10^6$ CFU/mL), adjusted to McFarland turbidity, was added to each dilution and incubated at 37 °C for 24 h for bacteria and 30 °C for 72 h for fungi. The lowest dilution with undetectable bacterial growth was recorded as the MIC. A lack of growth of the microorganisms was validated by the absence of turbidity after inoculating into agar followed by incubations for 24 and 72 h, for bacteria and fungi, respectively [62].

3.5. In Vitro Antiproliferative Activity Testing

All eight of the isolated compounds were tested against three cell lines, namely, human breast cancer cell line (HCC 1395), human prostate cancer cell line (DU 145) and normal cell line (Vero E6). Seven concentrations of each compound in dimethyl sulfoxide (DMSO) (100, 33.33, 11.11, 3.70, 1.23, 0.41 and 0.14 µg/mL) were used in triplicate. The chemotherapeutic drug doxorubicin was used as positive control in the aforementioned concentrations, while cells in minimum essential media were used as negative control. Methyl-tetrazolium (MTT) assay was used to determine the antiproliferative activity.

The cells preserved in liquid nitrogen in vials were removed and thawed in a water bath at 37 °C. This was followed by centrifuging the vial contents and transferring the supernatant into a growth minimum essential media (MEM) enriched with 10 v/v of Fetal Bovine Serum (FBS), 1 v/v antibiotic and 1 v/v L-Glutamine in a T75 culture flask, which was incubated for 48 h at 37 °C in 5 v/v CO₂ to reach confluence level. After cells for all the cell lines attained confluence, they were cleaned with saline phosphate buffer and harvested by trypsinization. The viable cells were calculated by determining cell density using the Trypan blue exclusion method in a haemocytometer. An aliquot of 100 µL containing 2.0×10^4 cells/mL suspension was seeded into a 96-well plate and incubated for 24 h at 37 °C in v/v CO₂. Fifteen microliters of test plant extracts were added at seven different concentrations to each serial dilution, from row H to B in a 96-well plate. Row A, containing cells and medium, only served as the negative control, while row H had the highest sample concentration (100 µg/mL). Doxorubicin was used as the positive control. The experiment was done in triplicates and incubated for 48 h. Thereafter, 10 µL of MTT dye (5 mg/mL) was added into each well and the plates were incubated at 37 °C for 2 h in 5% CO₂. The enzyme mitochondrial dehydrogenase, a biomarker of live cells, is known to interact with and reduce MTT dye to an insoluble formazan, which is purple in color. The amount of formazan formed is directly proportional to the number of live cells. Formazan formation was solubilized with 50 µL of DMSO and confirmed using an inverted light microscope. The antiproliferative activity of the test samples on the cancer and normal cells was expressed in IC₅₀ values (the sample concentration which killed 50% of the cancer cells) and CC₅₀ values (concentration of sample that exerted lethal effects on 50% of the normal cells) [63].

3.6. Data Analysis

3.6.1. Calculation of Percentage Cytotoxicity

Cytotoxicity was calculated using the following formula (Equation (1)) [64]:

$$\% \text{ Cell viability} = (A_T - A_B) / (A_C - A_B) \times 100 \quad (1)$$

where A_T = Absorbance of treated cells (drug), A_B = Absorbance of blank (only media) and A_C = Absorbance of control (untreated) (Equation (2)).

$$\% \text{Cytotoxicity} = 100 - \% \text{ cell viability} \quad (2)$$

The Finney's Probit analysis using BioStat version 6.7 was used to calculate the CC_{50} and IC_{50} values. The values were then analyzed using a Tukey's test to determine statistical significance at a probability level of 0.05.

3.6.2. Selectivity Index Determination

Selectivity index (SI), which shows the capability of a treatment to selectively exert toxicity towards cancerous cells and spare the normal cells, was determined using (Equation (3)) [65]:

$$SI = \frac{CC_{50}}{IC_{50}} \quad (3)$$

where CC_{50} is concentration of the compound that killed 50% of the normal cells, and IC_{50} is concentration of the compound killed 50% of cancerous cells.

4. Conclusions

In this study, eight compounds (myristic acid (1), stigmasterol (2), 8-acetyldihydrochelerythrine (4), arnottianamide (5), 10-methoxycanthin-6-one (6) and 8-oxochelerythrine (8)) were isolated and characterized from the root bark extract of *Z. paracanthum* for the first time, adding new information to better define the phytochemistry of this plant. Both the root bark extract and the purified compounds were tested using in vitro-based antimicrobial and antiproliferation cancer assays. All other previous investigations have examined only the stem extracts of this species. The present study thus supports and substantiates the work of others, as it highlights the value of analyzing different plant parts for bioactivity. Other studies of the chemistries and bioactivities of the leaf extracts of this species should be carried out, as leaves are a more renewable source of material for pharmaceutical application, compared to both stems and roots. This study provides scientific evidence for the popularity of root extracts of *Z. paracanthum* as a traditional plant medicine, which is taken for microbial infections and/or cancer-related ailments in locations where the plant is found.

Supplementary Materials: The following are available online at <http://www.mdpi.com/2223-7747/9/7/920/s1>, Table S1: 1H , ^{13}C NMR data and HMBC correlations for myristic acid (1), Table S2: 1H , ^{13}C NMR data and HMBC correlations for stigmasterol (2), Table S3: 1H , ^{13}C NMR data and HMBC correlations for sesamin (3). Table S4: 1H , ^{13}C NMR data and HMBC correlations for 8-acetyldihydrochelerythrine (4). Table S5: 1H , ^{13}C NMR data and HMBC correlations for arnottianamide (5), Table S6: 1H , ^{13}C NMR data and HMBC correlations for 10-methoxycanthin-6-one (6), Table S7: 1H , ^{13}C NMR data and HMBC correlations for canthin-6-one (7), Table S8: 1H , ^{13}C NMR data and HMBC correlations for 8-oxochelerythrine (8), Figure S1: The 1H NMR spectrum of myristic acid (1) observed at 500 MHz for $CDCl_3$ solution at 25 °C. Assignment is given in Table S1, Figure S2: The ^{13}C NMR spectrum of myristic acid (1) observed at 125 MHz for $CDCl_3$ solution at 25 °C. Assignment is given in Table S1, Figure S3: The 1H - 1H COSY spectrum of myristic acid (1) observed at 500 MHz for $CDCl_3$ solution at 25 °C, Figure S4: The 1H - ^{13}C HSQC NMR spectrum of myristic acid (1) observed at 500 and 125 MHz for $CDCl_3$ solution at 25 °C, Figure S5: The 1H - ^{13}C HMBC NMR spectrum of myristic acid (1) observed at 500 and 125 MHz for $CDCl_3$ solution at 25 °C. Assignment is given in Table S1, Figure S6: The ESIMS spectrum for myristic acid (1), Figure S7: The 1H NMR spectrum of stigmasterol (2) observed at 500 MHz for $CDCl_3$ solution at 25 °C. Assignment is given in Table S2, Figure S8: The ^{13}C NMR spectrum of stigmasterol (2) observed at 125 MHz for $CDCl_3$ solution at 25 °C. Assignment is given in Table S2, Figure S9: The 1H - 1H COSY spectrum of stigmasterol (2) observed at 500 MHz for $CDCl_3$ solution at 25 °C, Figure S10: The 1H - ^{13}C HSQC NMR spectrum of stigmasterol (2) observed at 500 and 125 MHz for $CDCl_3$ solution at 25 °C, Figure S11: The 1H - ^{13}C HMBC NMR spectrum of stigmasterol (2) observed at 500 and 125 MHz for $CDCl_3$ solution at 25 °C. Assignment is given in Table S2, Figure S12: The ESIMS spectrum of stigmasterol (2), Figure S13: The 1H NMR spectrum of sesamin (3) observed at 500 MHz for $CDCl_3$ solution at 25 °C. Assignment is given in Table S3, Figure S14: The ^{13}C NMR spectrum of sesamin (3) observed at 125 MHz for $CDCl_3$ solution at 25 °C. Assignment is given in

Table 1. Assignment is given in Table S3, Figure S15: The ^1H - ^1H COSY spectrum of sesamin (3) observed at 500 MHz for CDCl_3 solution at 25 °C, Figure S16: The ^1H - ^{13}C HSQC NMR spectrum of sesamin (3) observed at 500 and 125 MHz for CDCl_3 solution at 25 °C, Figure S17: The ^1H - ^{13}C HMBC NMR spectrum of sesamin (3) observed at 500 and 125 MHz for CDCl_3 solution at 25 °C. Assignment is given in Table S3, Figure S18: The ESIMS spectrum of sesamin (3), Figure S19: The ^1H NMR spectrum of 8-acetyldihydrochelerythrine (4) observed at 500 MHz for CDCl_3 solution at 25 °C. Assignment is given in Table S4, Figure S20: The ^{13}C NMR spectrum of 8-acetyldihydrochelerythrine (4) observed at 125 MHz for CDCl_3 solution at 25 °C. Assignment is given in Table S4, Figure S21: The ^1H - ^1H COSY spectrum of 8-acetyldihydrochelerythrine (4) observed at 500 MHz for CDCl_3 solution at 25 °C, Figure S22: The ^1H - ^{13}C HSQC NMR spectrum of 8-acetyldihydrochelerythrine (4) observed at 500 and 125 MHz for CDCl_3 solution at 25 °C, Figure S23: The ^1H - ^{13}C HMBC NMR spectrum of 8-acetyldihydrochelerythrine (4) observed at 500 and 125 MHz for CDCl_3 solution at 25 °C. Assignment is given in Table S4, Figure S25: The ^1H NMR spectrum of arnottianamide (5) observed at 500 MHz for CD_2Cl_2 solution at 25 °C. Assignment is given in Table S5, Figure S26: The ^{13}C NMR spectrum of arnottianamide (5) observed at 125 MHz for CD_2Cl_2 solution at 25 °C. Assignment is given in Table S5, Figure S27: The ^1H - ^1H COSY spectrum of arnottianamide (5) observed at 500 MHz for CD_2Cl_2 solution at 25 °C, Figure S28: The ^1H - ^{13}C HSQC NMR spectrum of arnottianamide (5) observed at 500 and 125 MHz for CD_2Cl_2 solution at 25 °C, Figure S29: The ^1H - ^{13}C HMBC NMR spectrum of arnottianamide (5) observed at 500 and 125 MHz for CD_2Cl_2 solution at 25 °C. Assignment is given in Table S5, Figure S30: The ESIMS spectrum for arnottianamide (5), Figure S31: The ^1H NMR spectrum of 10-methoxycanthin-6-one (6) observed at 500 MHz for CD_2Cl_2 solution at 25 °C. Assignment is given in Table S6, Figure S32: The ^{13}C NMR spectrum of 10-methoxycanthin-6-one (6) observed at 125 MHz for CD_2Cl_2 solution at 25 °C. Assignment is given in Table S6, Figure S33: The ^1H - ^1H COSY spectrum of 10-methoxycanthin-6-one (6) observed at 500 MHz for CD_2Cl_2 solution at 25 °C, Figure S34: The ^1H - ^{13}C HSQC NMR spectrum of 10-methoxycanthin-6-one (6) observed at 500 and 125 MHz for CD_2Cl_2 solution at 25 °C, Figure S35: The ^1H - ^{13}C HMBC NMR spectrum of 10-methoxycanthin-6-one (6) observed at 500 and 125 MHz for CD_2Cl_2 solution at 25 °C. Assignment is given in Table S6, Figure S36: The ESIMS spectrum for 10-methoxycanthin-6-one (6), Figure S37: The ^1H NMR spectrum of canthin-6-one (7) observed at 500 MHz for CD_2Cl_2 solution at 25 °C. Assignment is given in Table S7, Figure S38: The ^{13}C NMR spectrum of canthin-6-one (7) observed at 125 MHz for CD_2Cl_2 solution at 25 °C. Assignment is given in Table S7, Figure S39: The ^1H - ^1H COSY spectrum of canthin-6-one (7) observed at 500 MHz for CD_2Cl_2 solution at 25 °C, Figure S40: The ^1H - ^{13}C HSQC NMR spectrum of canthin-6-one (7) observed at 500 and 125 MHz for CD_2Cl_2 solution at 25 °C, Figure S41: The ^1H - ^{13}C HMBC NMR spectrum of canthin-6-one (7) observed at 500 and 125 MHz for CD_2Cl_2 solution at 25 °C. Assignment is given in Table S7, Figure S42: The ESIMS spectrum of canthin-6-one (7), Figure S43: The ^1H NMR spectrum of 8-oxochelerythrine (8) observed at 500 MHz for CDCl_3 solution at 25 °C. Assignment is given in Table S8, Figure S44: The ^{13}C NMR spectrum of 8-oxochelerythrine (8) observed at 125 MHz for CDCl_3 solution at 25 °C. Assignment is given in Table S8, Figure S45: The ^1H - ^1H COSY spectrum of 8-oxochelerythrine (8) observed at 500 MHz for CDCl_3 solution at 25 °C, Figure S46: The ^1H - ^{13}C HSQC NMR spectrum of 8-oxochelerythrine (8) observed at 500 and 125 MHz for CDCl_3 solution at 25 °C, Figure S47: The ^1H - ^{13}C HMBC NMR spectrum of 8-oxochelerythrine (8) observed at 500 and 125 MHz for CDCl_3 solution at 25 °C. Assignment is given in Table S8, Figure S48: The ESIMS spectrum for 8-oxochelerythrine (8).

Author Contributions: M.M.K. was involved in conceptual design of the project, data collection, analysis and wrote the first draft of the manuscript. C.W.L. and A.Y. were also involved in the conceptualization of this study and provided guidance on antimicrobial and anticancer assays throughout the course of the study. Parts of this work were completed in the laboratories of N.P.M., where some of the compound structure determination was conducted. N.P.M. was involved in data interpretation and provided editorial changes to the first manuscript draft. S.Y. was involved in the phytochemical isolation and structural determination of tested compounds with M.M.K., under the supervision of A.Y. J.G. made intellectual inputs and contributed towards funding parts of this study. All authors have read and approved the final manuscript.

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