

Experimental

The experiments were carried out with three-weeks grown callus cultures of five plant species, namely: *Leuzea carthamoides* DC. Asteraceae (1 mg 2,4 D + 1 mg K), *Bergenia crassifolia* (L.) Fritsch Saxifragaceae (10 mg IAA + 1 mg K), *Leonurus cardiaca* L. Lamiaceae (1 mg IAA + 1 mg K), *Rhodiola rosea* L. Crassulaceae (1 mg 2,4 D) and *Datura meteloides* DC. ex Don! Solanaceae (1 mg 2,4 D); out of the species mentioned above only *Bergenia* produces arbutin in intact plant. A 20-g aliquot of a raw callus (approx. 1 g of dried mass; size of the swelled callus particles 1–5 mm) was suspended in 300 ml of air-purged M-S medium (+ corresponding stimulators) doped with HQ (starting concentration of HQ in the medium was $0.85 \text{ mmol} \cdot \text{l}^{-1}$). The biotransformation changes in the medium were monitored at $23 \pm 2^\circ\text{C}$ for 48 h, the measurements of HQ concentration being performed in 30 min intervals with use of the automated amperometric FIA setup (Fig. 2). Selective FIA assay of 2 μM to 20 mM HQ in M-S medium was carried out with the use of a three-electrode flow-through amperometric cell of the wall-jet type with 0.1 M acetate buffer of pH 4.6 as the carrier stream; working electrode spectrographic graphite rod (diam. 3 mm) impregnated with epoxide resin (working potential $E_w = +0.5 \text{ V}$ vs. reference SCE); auxiliary platinum wire electrode. The overall content of arbutin in the spent calluses was determined by HPLC [2]. The biotransformation experiments were performed in duplicate; the results of the parallel runs were practically identical.

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Coumarins from *Hypericum keniense* (Guttiferae)

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Hypericum keniense Sweinf. (Guttiferae), a shrub or small tree found growing in rain forests in the tropical East Africa, is a hitherto phytochemically uninvestigated species [1]. Guttiferae plant species are widely used in folk

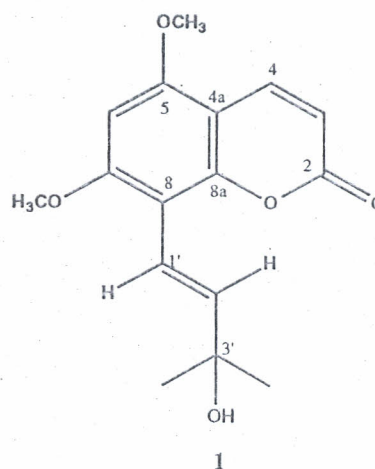


Table: ¹H NMR and ¹³C NMR data assignments and relevant HMBC correlations of (E)-8-(3'-hydroxy-3'-methyl-1'-butenyl)-5,7-dimethoxycoumarin (1)

Atom	¹³ C 75 MHz, CDCl ₃	¹ H 300 MHz, CDCl ₃	Relevant HMBC's
2	161.1		
3	110.8	6.16, d, J = 9.6 Hz	C2, C4a
4	138.7	7.98, d, J = 9.6 Hz	C5
4a	103.6		
5	155.6		
6	90.2	6.32, s	
7	161.0		
8	106.4		
8a	153.4		
5-OCH ₃	56.0	3.96, s	C5
7-OCH ₃	55.9	3.94, s	C7
1'	114.2	6.86, d, J = 16.5 Hz	C8, C7, C8a, C3'
2'	141.8	6.90, d, J = 16.5 Hz	C4', C5'
3'	77.6		
3'OH		2.02, s	
4'	30.0	1.46, s	
5'	30.0	1.46, s	

medicine and prior investigations into some of the species of this family led to the isolation of antiviral [2], antimicrobial [3, 4], antifungal [5] and cytotoxic [6–8] bioactive compounds including coumarins. In the present study, repeated chromatographic fractionation of the *n*-hexane and ethylacetate extracts of *H. keniense* stem bark afforded 5,7-dimethoxy-8-(3'-methylbut-2'-enyl)-coumarin (15 mg, 0.0010%), 8-(3',3'-dimethoxyxiranyl-methyl)-5,7-dimethoxy-chromen-2-one (71 mg, 0.0046%), toddanolactone (12 mg, 0.0008%), pimpinellin (62 mg, 0.0040%), the novel coumarin **1** and betulinic acid 756 mg (0.0484%). The chemical identities of these coumarins being reported for the first time from *H. keniense*, were established by comparing their physical and spectral data with those in the literature [9–12]. Compound **1** showed the elemental composition C₁₆H₁₈O₅ and was identified as (E)-8-(3'-Hydroxy-3'-methyl-1'-butenyl)-5,7-dimethoxycoumarin (5-methoxymurraol) from its ¹H NMR and ¹³C NMR (Table 1) together with the MS, UV and IR spectroscopic data. To our knowledge, this is the first report on this position 8 substituted tertiary allylic alcohol of a 5,7-dimethoxy coumarin from a natural source. Compound **1** can be regarded as a biosynthetic intermediate of the naturally occurring 5,7-dimethoxycoumarins gleinene and gleinadiene [12].